

Name:

Introduction to Earth Science NOTES

OBJECTIVES

Correctly define: observation, inference, classification, percent deviation, density, rate of change, cyclic change, dynamic equilibrium, interface, mass, volume

GRAPHICAL RELATIONSHIPS

- Graphically demonstrate the difference between direct, inverse, cyclic, and non- relationships.
- Classify events as cyclic or non-cyclic.
- Associate the words “cyclic” and “predictable”.

OBSERVATIONS, INFERENCES AND CLASSIFICATION

- Give an example of an observation.
- Develop an inference based on collected data.
- Classify objects based on their similarities or differences.

DENSITY, MASS, VOLUME

- Explain how density, mass and volume change as an object is heated, cooled or split apart.
- Calculate the density of regularly shaped and irregularly-shaped objects.
- Graphically show the relationship between temperature and density for: (1) water (2) all other objects
- State the temperature at which water is most dense and whether a object will sink or float in water based on its density.
- State the phase of matter in which most materials are most and least dense.

SCIENTIFIC INSTRUMENTS

- State the most common instruments used to measure the: (1) volume of regular, rectangular objects, (2) volume of irregularly shaped objects, (3) mass of objects, (4) distance between objects, and (5) time.

REAL-WORLD EXAMPLES

- Give real-life examples of a system in dynamic equilibrium, an interface, and three cyclic events.

FORMULAS AND WORD PROBLEMS

- Mathematically solve volume, density, percent deviation, and rate of change problems using the Earth Science Reference Tables.

Vocabulary

Classification:

Cyclic Change:

Density:

Dynamic Equilibrium:

Inference:

Interface:

Mass:

Observation:

Percent deviation:

Rate of Change:

Volume:

Key Concepts & Questions

Scientific Instruments

	COMMON INSTRUMENT
VOLUME OF REGULAR, RECTANGULAR OBJECTS (A BOX)	
VOLUME OF IRREGULARLY SHAPED OBJECTS (ROCKS)	
MASS	
DISTANCE	
TIME	

Formulas

Using the Earth Science Reference Tables, write the formula for each of the items below:

density:

percent error (deviation):

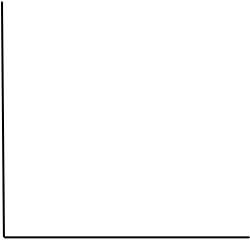
rate of change:

Graphical Relationships

A. Direct Relationship.

As one variable increases, the other _____.

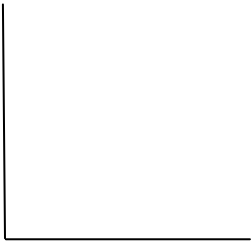
Examples:



B. Inverse Relationship

As one variable increases, the other _____.

Examples:

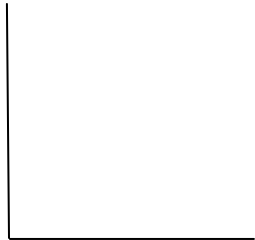


C. Cyclic Relationship

As one variable increases, the other _____

Examples:

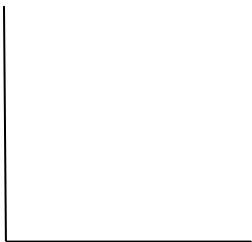
*Events that are cyclic are also _____ !



D. No relationship.

As one variable increases, the other _____.

Examples:



Observations, Inferences, Classification

What is used to make an observation?

After observations have been collected. What does it mean to make an inference?

Give examples of how scientists use classification systems.

DENSITY, MASS, VOLUME

MASS:

Name the common scientific instrument used to measure mass: _____

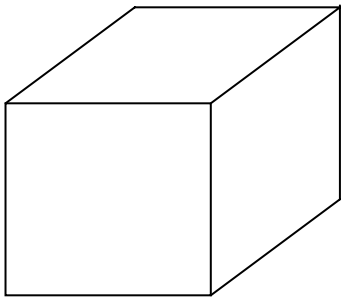
If an object is heated, what happens to its mass? _____

Why? _____

If an object has a mass of 240g on Earth, its mass on the moon will be (more, less, the same).

Why? _____

VOLUME of a regular rectangular object:



What instrument would be used to measure this object's volume? _____

What is the formula for finding the volume of this object?

Calculate the volume of this object to the nearest tenth of a centimeter. Show all formulas.

VOLUME of an irregularly shaped object:

What instrument would a student use to measure the volume of an object such as a rock?

Describe the process you would use.

DENSITY of All Objects EXCEPT WATER:

DENSITY = HOW TIGHTLY PACKED THE ATOMS ARE

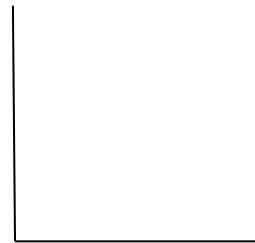
When an object is heated, it _____ and the atoms become (more, less) packed. Therefore the object becomes (more, less) dense.

When an object is cooled, it _____ and the atoms become (more, less) packed. Therefore the object becomes (more, less) dense.

What happens to the density of an object when it is split into smaller parts?

Why?

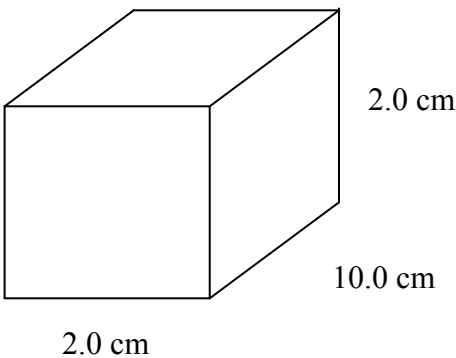
What is the formula for density?



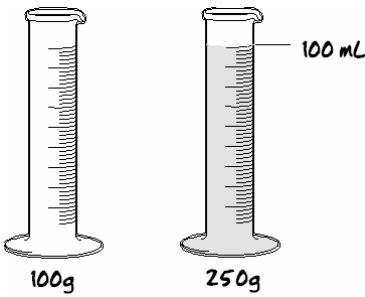
Sample problems:

1. A rock has a mass of 240g and a volume of 12cm³. Showing all formulas and calculations, determine the density of the rock.

2. The box below has a mass of 120g. Showing all formulas and calculations, determine the density of the box.



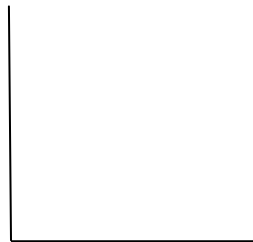
3. If the empty container has a mass of 100g and the filled container has a mass of 250g. What is the density of the liquid inside? Show all work below.



DENSITY OF WATER:

Water is most dense at _____°C. This is because water _____ above and below this temperature.

Draw the graph showing the density of water versus the temperature of water.



The density of water when it is most dense is: _____ g/mL

FLOAT OR SINK?

Any material with a density greater than water will _____.

Any material with a density less than water will _____.

Example. If an object has a mass of 25g and a volume of 50mL, will it sink or float in liquid water? Why?

PHASES OF MATTER AND DENSITY

During which phase of matter (solid, liquid or gas) are most materials:

most dense? _____

least dense? _____

Dynamic Equilibrium

Give a real life, earth science example of a system that is in dynamic equilibrium.

Interfaces

Give a real-life, earth science example of an interface.

Cyclic Events

Give three, real-life, earth science examples of cyclic events.